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Abstract: - Control of Chemical Plant is important because it further comprises of small plants ranging from water control up to the chemical composition control. Most of the chemical processes are non linear and non stationary. Also the process dynamics of these processes is not well understood due to some unknown disturbances. So fuzzy logic control is best suited for such types of processes where process dynamics is complex. In present paper, a Fuzzy logic system has been purposed that is particularly suited for Sulfuric acid plant. The system design starts from identification of inputs, Outputs and choosing the membership function for each condition from normal operation up to emergency operation of sulfuric plant. A complete automation and control of sulfuric acid production process has been implemented which show consistent results.

Keywords: - Fuzzification, Inference Engine, defuzzification, process dynamics, membership functions

I. Introduction

Most of chemical reaction processes are nonlinear in nature. Moreover its modeling is also complicated. In chemical industry, the processes are in general complex, with delays, non-linearity; it is not always possible to control them with classical regulators [1]. The major disadvantage in fuzzy control is lacking analytical design technique i.e. determination of parameters of Membership functions [2-3]. Basic knowledge of pure physics and chemistry is needed with advanced engineering for the development of Instrumentation system to chemical process control. Conventional and advanced controllers (like PID & MRAC respectively) need most accurate modeling; in addition simulation has also some defects in virtual implementation and testing. Fuzzy controllers have a non-linear behavior that makes them a useful tool for the chemistry industry [4].

The application of the Fuzzy Set Theory to a wide range of control applications has made possible the establishment of intelligent control in these areas [5]. A relationship is to be developed between the inputs and outputs with the help of the fuzzy logic controller [6]. Sulfuric Acid production is also one of the nonlinear and in addition exothermic one. Here two parameters, concentration and temperature plays vital role and shows nonlinear behavior while producing sulfuric acid. For accurate and safe operation one must have a very good mathematical model of both plants as well process. It is difficult to get good control and accuracy due to less reliability of mathematical model or errors in the model. So instead of conventional or advanced controller which totally depends on the mathematical model, fuzzy logic control strategy has been selected as an option. This scheme of control will evaluate certainty within uncertainty and handling the parameters within range to control the entire process continuously though fluctuations occurs [7-8]. Fuzzy control provides effective solutions for nonlinear and partially unknown processes, mainly because of its ability to combine information from different sources [9]. The fuzzy knowledge base consists of a set of physically interpretable if- then rules providing physical insight into the process [10]. Therefore the process industries require more reliable, accurate, robust, efficient and flexible control systems for the operation of process plant.

Process dynamics: Production of sulfur dioxide II.

The flow diagram of sulfuric acid manufacturing process is shown in figure 1. Solid Sulfur is melted by steam coils at 140°C in brick lined tanks. The molten sulfur is filtered to remove any impurities (usually iron or organic compounds). The molten sulfur is pumped to the burner where it is burnt in an excess of dry air. The gas exiting the burner is maintained at 8-9% v/v sulfur dioxide and approximately 830°C due to the heat produced by the exothermic reaction [3]. This reaction is described by the equation [3]:

 $S + O_2 \rightarrow SO_2$ $\Delta H = -300 \text{ kJ mol}{-1}$ (1)

The sulfur dioxide/air gas mixture is then passed through the hot gas filter situated besides the air burner, where any ash contamination is removed. The sulfur dioxide is converted to sulfur trioxide by reacting with oxygen over a catalyst. This reaction is described by the equation:

> $SO_2 + \frac{1}{2}O_2 \rightarrow SO_3$ (2)

 $\Delta H = -100 \text{ kJ mol}-1$

This reaction occurs in the converter in which a four-stage reaction vessel, with each stage consisting of a solid catalyst bed through which the gas is passed. The catalyst used is vanadium pent oxide (V_2O_5) and potassium sulphate dispersed on a silica base; this forms a porous support, giving a large surface area for reaction. It is believed that the V_2O_5 increases the rate of the overall chemical reaction by oxidizing the SO₂ to SO₃ and

being re-oxidized itself by the oxygen in the gas stream. This reaction is exothermic and its equilibrium constant decreases with increasing temperature. Figure 1 shows the percentage conversion of SO_2 to SO_3 that would be reached at an SO_2 concentration of 8% v/v and a range of gas temperatures. However, the reaction rate is also temperature dependent, so that if the temperature becomes too low the equilibrium point will not be reached. In practice, the gas temperature must be maintained between 400 - 500°C to maintain a high reaction rate and also high conversion equilibrium [3]. As the reaction is exothermic, heat is generated across each of the catalyst beds. This heat must be removed between each stage to maintain the optimum reaction temperature into the following stage. The greatest degree of cooling is required between the first and second stages. Cooling after the second and third stages is by injection of dried air. The gas is passed to the absorption tower, a packed tower where SO_3 is absorbed into a counter-current flow of 98 - 99% sulfuric acid. The overall reaction can be described by the following equation, where sulfur trioxide reacts with the free water to produce sulfuric acid:



Figure 1: Flow Diagram of Sulfuric acid manufacturing process

 $\Delta H = -200 \text{ kJ mol}-1$

$$SO_3 + H_2O \rightarrow H_2SO_4$$
 (3)

The circulating sulfuric acid must be maintained at about 98% concentration and 70° C to maximize the absorption efficiency. The acid strength is important because the vapor pressure of sulfur trioxide above sulfuric acid is at a minimum at an acid strength of 98% which is depicted in figure 3. At higher concentrations the increased vapor pressure is caused by SO_3 and at lower concentrations the water vapor pressure increases sharply and the resultant acid mist is not readily re-absorbed and escapes to the atmosphere.



Figure 2: Effect of temperature on the conversion of SO_2 into SO_3 Figure 3: vapor pressure above sulfuric acid

The sulfuric acid is circulated at such a rate that there is only a very small increase in concentration through the absorber tower. Dilution water is added to the circulating acid tank and also as atmospheric water absorbed in the drying tower. A stream of sulfuric acid is continuously bled off and cooled through a plate heat exchanger before being passed into the storage tanks. The overall conversion from sulfur to sulfuric acid is greater than 98.5%. The plant operates under an air discharge permit which controls emissions of sulfur dioxide and total acidity. Traditionally mild steel has been used as the primary material of construction for process equipment containing 98% sulfuric acid. The corrosion rate is reasonably low, except at the air/liquid interface where atmospheric moisture encourages corrosion. Raw sulfur is melted and cleaned then it is burnt in the presence of air, here moister contamination prevention is the important criteria. Second is the production of SO₂ gas is also important criteria because volume flow and temperature of SO₂ gas will decide how the boiler would work, another important point over here is reactor's 1st and 2nd stage temperature which is to be maintained

according to figure shown above. The reactor temperature is important because according to temperature deviation in the all four stage the rate of conversion of sulfuric dioxide to trioxide also get affected. In 3rd step of the process plant the concentration plays a vital role for both sulfuric acids as well sulfur trioxide.

III. Fuzzy logic control

Fuzzy logic can be viewed as an extension of multi valued logic system. In two valued logic system a proposition is either true or false. In multi valued logic system a proposition may be true or false or have an intermediate truth value which may be an element of infinite truth valued set. In fuzzy logic, the truth values may range over the fuzzy subset. The fuzzy logic controller is designed to deal with the situations where available source of information are inaccurate, subjectively interpreted or uncertain. The main constitutes of fuzzy controller are Fuzzification, Knowledge base, Rule base, Inference strategy and defuzzification. Figure 4(a) shows the Block diagram of Fuzzy controller. Fuzzification is the process of converting the inputs variable values (sensor signal values) into linguistic variable values or membership function in fuzzy logic sets (fuzzy values). Rule-base of fuzzy logic breaks the control problem down into a series of IF X AND Y THEN Z rules that define the desired system output response for given system input conditions . The number and complexity of rules depends on the number of input parameters that are to be processed and the number of fuzzy variables associated with each parameter[7-8]. Create fuzzy logic membership functions that define the meaning (values) of input/output terms used in the rules. Defuzzification is the process in which output linguistic variable value (fuzzy variable value) is translated into crisp value (real value).



Figure 4(a): Block diagram of Fuzzy Controller

FUZZY TECH SOFTWARE 5.5 is used for the automatic control of sulfuric acid manufacturing process. It contains all the editors, analyzers and tools to design a complete fuzzy logic system. It supports various fuzzy logic inference methods and algorithms. The first step in a fuzzy logic system design is the definition of the system structure. Here, we define the inputs & outputs of the fuzzy logic system and how they interact. As shown in figure 4(b) the small blocks on the left side are the input interfaces. The input interfaces also contain the fuzzification of the input values. The icon on the left indicates the employed fuzzification method. The small blocks on the right side are the output interfaces that contain the defuzzification method. The larger block in the middle of the screen is the rule blocks. The rule blocks each contain an independent set of fuzzy logic rules. The left column shows the variables used in the precondition of fuzzy rules. The right column shows the variables used for the conclusion of fuzzy rules. The upper box displays the Condition Aggregation Operator.



Figure 4(b): Text block explaining object of Fuzzy control system

IV. Process Description for development of fuzzy control strategy:

The P&ID diagram of Sulfuric acid manufacturing process has been developed and presented in figure 5. The abbrevation of various instrument and physical variables used in automation and control of sulfuric acid plant has been presented in Table 1. Sulfur is feeded to the sulfur melting tank. Here it is melted with fix temperature steam. Tank's level and temperature would decides whether to feed and how much to feed solid sulfur and steam to the tank. The melted sulfur is pumped to the next tank called dirty sulfur tank. The solid sulfur, which is melted & converted to the liquid state, having some contamination needed to be removed. So this dirty sulfur is pumped & passed to the sulfur filter and then to the clean sulfur tank. The MTBM (Mean Time between Maintenance) means the sulfur filter chocking condition would be counted in advance. The pressure switch of sulfur filter will alert the control system when to replace the filter & between this times production would not be paused though earlier stage pumping would stop, this is due to independency of level parameter of clean sulfur tank.

Step 1: The clean sulfur is pumped to the air burner with respect to clean sulfur tank level, air burner pressure, air burner temperature & dryer humidity conditions, synchronously air from the dryer faded to the air burner according to sulfur flow to the burner, pressure & temperature of air burner & dryer humidity conditions. Dryer is feeded with air coming through filter & dryer volume is controlled through a pressure switch. Here when dryer humidity crosses its limit, this pause the whole SO₂ production process & further so on. But most of MTBM for changing dryer beds are so set that it wont affects the production process. On the basis of volume flow, temperature of produced Sulfuric Dioxide gas, temperature of boiler & pressure of boiler, the feed control valve of Sulfuric Dioxide gas to the boiler is controlled. The boiler is waste heat boiler which converts the waste heat produced due to exothermic reaction to the usable steam. The feed water to the waste heat boiler is controlled by the feed water control valve which is dependent of parameters are level, pressure & temperature of the boiler. In addition temperature of 3rd & 4th stage of four stage converter/reactor & humidity condition of dryer decides the air feeding to those particular stages of converter/reactor.

Step: 2 Sulfuric dioxide gas is converted into sulfuric trioxide gas inside the converter/reactor with the help of vanadium pant oxide catalyst bed. Three parameters, temperature of 1^{st} stage of converter/reactor, temperature and flow of the Sulfuric Dioxide gas coming from the boiler decides feed control to the converter/reactor through the control valve for reactor sulfuric dioxide gas feed control. Cooling of Sulfuric Trioxide gas is needed after passing through 1^{st} stage of reactor because having reaction with vanadium pent oxide (V₂O₅) sulfuric dioxide's temperature would again rise due to this reaction is also be an exothermic one. The coolant feeded to the heat exchanger for the cooling of this sulfuric trioxide gas is controlled with the coolant flow control valve which is operated on the conditions of the two temperature parameters; these are temperature of 1^{st} and 2^{nd} stage of reactor.

Step3: Concentration of sulfuric trioxide gas coming to two stage absorber, concentration of sulfuric acid, vapor pressure inside 1st stage of absorber & concentration of sulfuric trioxide gas venting from absorber would decide volume flow of sulfuric trioxide gas entering to a two stage absorber through sulfuric trioxide flow control valve. The unabsorbed sulfuric trioxide gas having less concentration is to be vented and gas having higher concentration is to be recycled, this is done by stopping the feed of sulfuric trioxide coming from reactor & by stopping vent from absorber with opening of recycle valve. This is done with the help of one three way valve and one recycle valve, one angle and one linear on off type. The operation of these valves would be decided by vapour pressure inside the absorber, concentration of sulfuric acid and concentration of sulfuric trioxide gas being vented from the absorber. The feed rate of dilution water to the high concentration sulfuric acid being stored in the acid circulation tank is decided by level of the acid circulation tank, concentration of sulfuric acid inside the acid circulation tank and flow of high concentration sulfuric acid towards the acid circulation tank. The circulation of lower concentration sulfuric acid is done with the help of circulation pump of which's speed is controlled through level of the acid circulation tank, concentration of sulfuric acid inside that tank and concentration of sulfuric acid of 2nd stage of absorber. The vapor pressure above the sulfuric acid inside the acid circulation tank is controlled through two valves. One valve feed additional air pressure when vapour pressure is lower than desired and one valve vents the vapour by opening when pressure is higher than desired. The pumping of sulfuric acid of desired concentration is pumped to the acid storage tank in accordance to level of both acid circulation and storage tanks and pumping condition of acid circulation pump. And the final output is taken on the basis of level of storage tank with storage tank pump in respect.



Figure 5: Process Equipment and Instrumentation lay out of the process

Faninments	Instruments
HP: Hoper for Sulfur Feed 0 to 10 kgmm	
SMT: Sulfur Melting Tank	Level All having 0 to 10 ft level
DST: Dirty Sulfur Tank	L1: SMT tank level
SF: Sulfur Filter	L2: DST tank level
CST: Clean Sulfur Tank	L3: CST tank level
AB: Air Burner with Ash Filter inside	L4: Boiler water level
DR: Dryer	L5: ACT tank level
AF: Air Filter	L6: AST tank level
WHB: West Heat Boiler	
4SCR: Four Stage Convertor/Reactor	Temperature
HE: Heat Exchanger	T0: SMT tank temp.120 to 160 °C
2SAR: Two Stage Absorber	T2: AB temp. 800 to 900 °C
ACT: Acid Circulation Tank	T3: SO2 line to boiler temp.800 to 900 °C
AST: Acid Storage Tank	T4: Boiler temp.700 to 900 °C
	T5: SO2 line to 4SCR temp.600 to 700 °C
Pumps All having 0 to 500 rpm speed	T6: 1" stage temp of 4SCR, 825 to 945 °C
P1: MST pump to DST	T7: 2 nd stage temp of 4SCR 625 to 725 °C
P2: DST pump to CST	T8: 3 rd stage temp of 4SCR450 to 550 °C
P3: CST pump to AB	T9: 4th stage temp of 4SCR400 to 450 °C
B1: Air Suction Draft for dryer	
P4: ACT to 2SAR pump	Switches
P5: ACT to AST pump	S1: SF pressure switch
P6: AST pump for final product out.	\$2: DR humidity switch
	S3: DR pressure switch
	S4: AF pressure switch
	CS1: Venting SO3 concentration switch30% = low & 60% = high
Concentration	Control Valves
C1: Concentration of SO3 from 4SCR 60% to 90%	V1: Steam for melting sulfur tank
C2: Concentration of Sulfuric acid in 1* stage of 2SAR 97% to 99%	V2: Air to burner control valve
C3: Concentration of Sulfuric acid in 2nd stage of	V3: SO2 flow control to boiler

Table 1: Abbreviation of various instrument and physical variables

2SAR 92% to 94%	
22	V4: Boiler feed water control
Pressure	V5: SO2 flow control to 4SCR
PT1: AB pressure 0 to 20 kgpcme	V6: Coolant flow control to HE
PT2: Boiler Pressure 30 to 50 kgpcm2	V7: Air control for 3 rd stage of 4SCR
PT3: Vapour pressure in ACT	V8: Air control for 4th stage of 4SCR
PT4: Xapour pressure in 2SAR	V9: SO3 flow control from 4SCR
	V10: recycle SO3 flow control for 2SAR
Flow	TV1: Three way valve for vent or recycle of SO3
F2: CST to AB flow of sulfur 0 to 10 CFPM	V11: Dilution water control to ACT
F3: SO2 gas flow to boiler 0 to 10 CFPM	V12: Air control for ACT
F4: SO2 gas flow to 4SCR 0 to 10 CFPM	V13: Vent control of sulfuric acid yspour.
F6: Sulfaric acid flow from 4SCR to ACT 0 to 10 CFPM	

4.1. Defining Control Loops and Application of Fuzzy Logic for step1

The complete fuzzy logic based control system for step 1 excluding the controller for control valves V3 and V4 has been illustrated in figure 6. and the summary of rules for all steps are illustrated in table 2.



Figure 6: Fuzzy Logic control for Step-1 excluding the control of valves V3 and V4

Table 2:	Fuzzy Kules For Step 2
L1 & TO THEN HP	F4 & T5 & T6 THEN V5
L1 & T0 & HP THEN V1	T7 & T6 THEN V6
L1 & L2 & P2 TERIN PT L2 & L3 & P3 & S1 THEN P2	For Step 3
PT1 & L3 & T2 & S2 THEN P3	C1 & C2 & CS1 & PT4 THEN V9
PT1 & F2 & T2 & S2 THEN V2	C2 & CS1 & PT4 THEN TV1
T8 & S2 THEN V7	15 & C5 & F6 THEN V11
T9 & S2 THEN V8	L5 & C5 & C3 THEN P4
F3 & T3 & PT2 & T4 THEN V3	L5 & PT3 THEN V12 THEN V13
L4 & P12 & 14 THEN V4	L5 & L6 THEN P5 L6 THEN P6

(i) HP hoper controls the quantity of solid sulfur being fed into the SMT. Control parameters for SMT are level L1 and temperature T0. Here V1 is the control valve which feeds fix temperature steam to the tank for melting the solid sulfur. So we can define the control Loops over here. The membership function of each **are** defined in figures 7-10. IF L1 & T0 THEN HP; IF L1 & T0 & HP THEN V1



Figure 7: Member ship function of L1







Figure 8: Member ship function of T0





(ii) Pumping of sulfur is done by three pumps are P1, P2 & P3. P1 is controlled through parameters L1, L2 & pump P2 speed conditions. While P2 is controlled through parameters L2, L3, pump P3 speed conditions & sulfur filter switch S1's conditions. Pump P3 is functioning for sulfur feeding to air burner so it is pumped parameters L3, T2, PT1 & dryer humidity switch conditions. The membership function of each **are defined** in figures 11-19. IF L1 & L2 & P2 THEN P1; IF L3 & L2 & P3 & S1 THEN P2; IF L3 & PT1 & T2 & S2 THEN P3





Figure 12: MSF for P2



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Figure 16: MSF for P3



Figure 18: MSF for T2













(iii) Blower B1 feeds air to dryer and dryer feeds air to Burner via Air to Burner valve V2. B1 depends on humidity S2, Air filter pressure S4 and Dryer pressure S3. While V2 will be operated with sulfur flow F2, Air Burner Pressure PT1 & Temperature T2 and dryer humidity switch S2. The membership function of each **are defined** in figures 20-28. IF PT1 & F2 & T2 & S2 THEN V2; IF S4 & S3 & S2 THEN B1; IF S2 & T8 THEN V7; IF S2 & T9 THEN V8



Figure 20: MSF for F2







Figure 23: MSF for S4



Figure 24: MSF for B1

(iv) 3^{rd} Stage temprature cooling vlave (V7) control of 4SCR





Figure 26:MSF for V7







4.1.1. Fuzzy Logic Design for Step-1 for controlling control valve V3 & V4

(i) Sulfuric Dioxide gas coming from AB will be controlled by its flow F3 & temperature T3 with Boiler's Pressure PT2 & temperature T4. The membership function of each **are defined** in figures 29-33. IF F3 & PT2 & T3 & T4 THEN V3





















(ii) Feed water for Waste Heat Boiler is controlled through V4 valve with L4, PT2 & T4. The membership function of each **are defined** in figures 34-35. IF L4 & PT2 & T4 THEN V4



Figure 34: MSF for L4

Figure 35: MSF for V4

The complete Fuzzy Logic Control for Step 1 for control of valves V3 & V4 has been illustrated in figure 36.



Figure 36: Fuzzy Logic Control for Step 1for control of valves V3 & V4

4.2. Defining Control Loops and Application of Fuzzy Logic for step2

The complete fuzzy logic based control system for step 2 has been illustrated in figure 37 and the summary of rules for all steps are illustrated in table2.



Figure 37: Fuzzy Logic Control for Step 2

(i) Sulfuric dioxide gas from boiler is controlled by its flow F4 & temperature T5 and temperature of first stage of reactor T6. The membership function of each **are defined** in figures 38-41. IF F4 & T5 & T6 THEN V5









Figure 40: MSF for T6



(ii) The coolant to the Heat Exchanger flow control is done by valve V6 which is dependent on 1st and 2nd stage temperature of reactor. The membership function of each **are defined** in figures 41-42. IF T6 & T7 THEN V6





Figure 42: MSF for V6

4.3.1 Defining Control Loops and Application of Fuzzy Logic for step3: stage1

(i) Feed Control Valve(V9) for Absorber : Concentration of SO3 from reactor and concentration of H_2SO_4 & vapour pressure inside absorber with venting SO₃ concentration are parameters to control V9 valve. The membership function of each **are defined** in figures 43-47. IF C1 & C2 & CS1 & PT4 THEN V9













Figure 46: MSF for PT4



Figure 47: MSF for V9 www.iosrjen.org



The fuzzy project editor for Feed Control Valve(V9) for Absorber is illustared in figure49.



(ii) Fuzzy Control System for Absorber feed and Vent control

Low concentration and unabsorbed Sulfuric Trioxide gas should be vent in the atmosphere and higher concentration should be recycled. This is done by one three way valve and one on-off type valve. The membership function of each **are defined** in figures 50-51. IF C2 & CS1 & PT4 THEN TV1 THEN V10



Figure 50: MSF for TV1

Figure 51: MSF for V10

The fuzzy project editor for Absorber feed and Vent control is illustated in figure 52.



Figure 52: Fuzzy Control System for Step 3 stage one

4.3.2. Defining Control Loops and Application of Fuzzy Logic for step3: stage2

(i) Concentration of sulfuric acid inside circulation tank and tanks level and flow of sulfuric acid towards tank are the parameters to control V11. The membership function of each **are defined** in figures 53-56. IF C5 & L5 & F6 THEN V11



Figure 53: MSF for C5





Figure 55: MSF for F6

Figure 56: MSF for V11

(ii) Pump for acid circulation is controlled by the circulation tank's concentration and level with concentration of sulfuric acid inside 2^{nd} stage of absorber. The membership function of each **are defined** in figures 57-58. IF C3 & C5 & L5 THEN P4



Figure 57: MSF for C3

Figure 58: MSF for P4

(iii) For Vapour Pressure control inside the Acid Circulation Tank Two valves used, one (V13) is for venting vapour fume with high pressure and by other (V12) inserting air and generating pressure. The membership function of each **are defined** in figures 59-61. IF L5 & PT3 THEN V12 THEN V13





Figure 60: MSF for V12



Figure 61: MSF for V13

(iv) Storage Tank's pump runs on level of circulating and storage tank's level and circulating pump's speed condition. The membership function of each **are defined** in figures 62-64. IF L5 & L6 & P4 THEN P5





Figure 63: MSF for L6



Figure 64: MSF for P4

(v) Storage Tank gives the output through its pump according to level of storage tank. The membership function of P6 is defined in figures 65. IF L6 THEN P6



Figure 65: MSF for P6

The complete fuzzy logic control for Acid circulation and storage and out of step 3 is illustrated in figure 66.



Figure 66: Fuzzy Control System for Step 3 stage 2

V. Results

The Fuzzy logic system made for step 1 is controlling the production of Sulfuric Dioxide gas with high temperature. Here the temperature and volume both are controlled. In addition if the limits exceeded then the production would stop and the same for the moister contamination would also be controlled. Different control strategies are fulfilled i.e control achieved for the sulfur melting, control achieved for pumping of sulfur and cleaning of sulfur and control achieved for air burner to producing sulfuric dioxide gas which is depicted by Time plot for HP in **figure 67**. In second control system, boiler and feed to the boiler of both water and sulfuric dioxide gas are controlled and on reaching the limit the feed of both would stop. But this also would increase the Air Burner's temperature and pressure which will be controlled by the first controller. The results are shown in figure 68 which shows that control is achieved for this particular process.



Figure 67: Time Plot for HP

Figure 68: Time Plot for V3

The third control system controls feed to reactor and feed of coolant to heat exchanger respectively. Figure 69 shows time plot for the control achieved in the particular process. The time plot shows that control is achieved for this particular process. Fourth & fifth control system controls feed of sulfuric trioxide to the absorber, vent of residue sulfuric trioxide to the environment, concentration control of both stages of absorber (by feeding sulfuric trioxide gas to first stage and by feeding low concentration sulfuric acid to the second stage this is what the pumping of sulfuric acid through the absorber) and storage of ready to dispatch sulfuric acid. The time plot in figure 70 shows that control achieved in.







VI. Conclusions

Five fuzzy control systems for the entire plant have been designed with loop identification, which is showing consistent results as elaborate in previous **sections**. If this work would be done by conventional method then it would require controller for each loop, thus fuzzy control system reduces both modeling for the process-plant and detailed analysis for each loop for conventional control system. In addition expanded span with the help of linguistic variables give better approach for production with least errors; this shows reliability for this control phenomenon. The approach which circumvents the need for elaborating mathematical model of the complex chemical process and allows the use of qualitative information as linguistic rules is making knowledge base by integrating all individual fuzzy control loops and making a single control system. Mass Production of Sulfuric Acid from the higher capacity plant require to control number of melting tanks, air burners, west heat boilers with reactors and absorber. Neural fuzzy control may provide better solution for the control of large process plant.

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